

Optimization & Comparative Study of Microparticulate System of Flurbiprofen Prepared by Different Methods

Prashant Patil ^{1*}, Santosh Singh², Girish kashid³

Abstract

Comparative study of ionic gelation & spray dry methods for preparation of microspheres to deliver drugs has many other advantages over one another, like control-release of the drug, increase bioavailability and target delivery of the drug to the required site. In this research work shows the use of encapsulating sodium alginate, sodium carboxy methyl cellulose in biodegradable microsphere delivery system, to be delivered orally via a capsule, to gives desired therapeutic action. Microsphere formulations have merits over conventional tablet or capsule formulations, since it increases the surface area exposed to the absorption site and thus increasing the absorption of the drug and decreasing the dosing frequency of the drug. Flurbiprofen is a NSAIDs which is used widely in different intestinal diseases ulcerative colitis, Crohn's disease, carcinomas and infections. Flurbiprofen shows maximum absorption in the lower gastrointestinal tract regions, and shows half-life 4 hrs, it shows low bioavailability orally. The microsphere formulations were evaluated for its production yields, actual drug content, encapsulation efficiency, % Swelling Index release study is done by in vitro release analysis, mucoadhesive strength determination in vitro & in vivo methods.

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Introduction

The idea of mucoadhesive system came from the need to localize drug at a certain site in the body, often as the extent of drug absorption is limited due to the residence time of drug at the absorption site. In oral drug delivery, drug absorption is limited due to the gastrointestinal transit time of the dosage form. To illustrate suppose if a drug dosage form is to deliver a drug in a sustained manner for treating some chronic disease then it is required that the dosage form should remain at the site of absorption which is mainly upper part of the intestine, for a prolonged period of time but this is limited due to the gastrointestinal transit of the dosage form, so mucoadhesive dosage forms are formulated with the purpose of binding with the mucus layer of the GIT and thus increasing the residence time of the drug and also providing the intimate contact between a dosage form and the absorbing tissue and hence enhancing the absorption of the drug1-3.

Materials and Methods

Flurbiprofen gift sample available from Teva Pharma (Pvt.) Ltd., while Sodium alginate & Sodium carboxy methyl cellulose (Colorcon Ltd., UK). Central composite design was successfully applied from Design Expert software, version 7.0.0, State-Ease, Inc., Minneapolis. Microsoft Excel, DD solver and SPSS 17.0 (SPSS Inc) were used for the assessment of drug release data. Micropsheres formulations were evaluated for release study, percentage yield, actual drug content and encapsulation efficiency.

Microsphere Preparation method

Ionic gelation4-8:

1) Sodium alginate was added to mucoadhesive polymer & dissolved in purified

water forms homogenous polymer solution.

Corresponding author: Prashant Patil

Address: 1,2 Department of Pharmaceutics, School of Pharmacy Suresh Gyan Vihar University Jaipur - 302017, Rajasthan, India,



2) Drug add to polymer alginate mixture stirred to form clear solution resulted solution

was then added drop wise into 5% calcium chloride solution by syringe.

3) Added droplets were retained in calcium chloride solution for 25mins to complete

reaction produce spherical & rigid microspheres. 4) Product wash with water & dried 450c for 12 h. Formulation composition given in Table.4

Factorial Batches

А 32 factorial design was implanted for optimization of oral controlled release tablet

Combination Batches for microspheres

³Department of Pharmaceutical chemistrys School of Pharmacy Suresh Gyan Vihar University Jaipur - 302017, Rajasthan, India tablets. According to the model it contains two independent variables at three levels +1.0 and -1 (Table.2). According to the model total nine formulations possible. The composition of different formulations is shown in (Table.4). The different independent variables include: amount of HPMC K4 M (X1) & amount of sodium alginate (X2). Where HPMC K4 M (X1) & sodium alginate act as a and controlled release polymers. The different dependent responses include: % drug release at 8 hour(Y1), Time taken to release 50% drug, T50% (Y2), Time taken to release 90% drug, (Y3).

Batch Codo	Variable levels in	n Coded form
Datch Coue	X1	X2
F1	+1	+1
F2	+1	0
F3	+1	-1
F4	0	+1
F5	0	0
F6	0	-1
F7	-1	+1
F8	-1	0
F9	-1	-1

Table1: Factorial Design for Preparation of Batches

Translation of coded levels in actual units

Table2: Factors and their corresponding levels for the construction of 32 factorial design

Variable levels	Low (-1)	Medium (0)	High (+1)
X1= Concentration of	60	80	100
HPMC K4 M (mg)			
X2= Concentration of	60	80	100
sodium alginate (mg)			

Factorial formulations

Table3: Combination batches by using HPMC K4M & sodium alginate in various concentrations according to 32 factorial design.

Batch code/Content (mg)	F1	F2	F3	F4	F5	F6	F7	F8	F9
Flurbiprofen	100	100	100	100	100	100	100	100	100
HPMC K15M	100	100	100	80	80	80	60	60	60
sodium alginate	60	80	100	60	80	100	60	80	100
Microcrystalline cellulose	80	60	40	60	40	20	40	20	00
Total	300	300	300	300	300	300	300	300	300



Spray dray technique9-13

Mucoadhesive microspheres were prepared by spray drying technique. An aqueous solution containing different combinations/ratios of the polymers (Table 5) were prepared by dissolving sodium alginate and carboxy methylcellulose in distilled deionized water. The drug (1 g), previously dissolved in 100 ml of absolute methanol, was added to the polymer solution and sonicated using Ultra sonicator (1204 AU-Vibracell, USA) to obtain a homogeneous mixture. A crosslinking agent, glutaraldehyde (0 – 0.30 ml), was added to the homogenized solution and The resultant solution was spray dried using LU-222 ADVANCED lab spray drier (Labultima, India) for preparing microspheres through the nozzle of a spray-dryer (JISL, LSD- 48 mini spray dryer, India) at input temperature of 115 -117 °C, output temperature of 80 – 85 °C at 2 % feed rate and vacuum pressure of 35 psi (2.4 kg/cm2). The resulting microspheres were collected from the spray dryer and kept in a desiccator containing silica gel pending further tests.

Factorial formulations

CONTENT	F1	F2	F3	F4	F5	F6	F7	F8	F9
(Wt in mg.)									
Flurbiprofen : Sodium	1:2:0	1:2:1	1:2:2	1:2:0	1:2:1	1:2:2	1:2:0	1:2:1	1:2:2
alginate : Sodium CMC									
Cross linking agent (%)	00	00	00	20	20	20	30	30	30

Table4: Combination batches by using Sodium alginate & CMC in various concentrations according

Factorial Batches10-13

A 32 factorial design was implanted for optimization of oral controlled release tablet tablets. According to the model it contains two independent variables at three levels +1,0 and -1. According to the model total nine formulations possible. The composition of different formulations is shown in (Table.5). The different independent

variables include: drug: polymer ratio (X1) &percentage of Cross linking agent (X2), Where carboxyl methyl cellulose & sodium alginate act as an controlled release polymers. The different dependent responses include: % drug release at 8 hour (Y1), Time taken to release 50% drug, T50% (Y2), Time taken to release 90% drug, (Y3).

Combination Batches for microspheres

Table5: Factorial Design for Preparation of Batches

Batch	Variable levels in Coded				
	form				
Coue	X1	X2			
F1	+1	+1			
F2	+1	0			
F3	+1	-1			
F4	0	+1			
F5	0	0			
F6	0	-1			
F7	-1	+1			
F8	-1	0			
F9	-1	-1			

X1 : drug: polymer ratio X2 : Concentration of Cross linking agent



Evaluations of Microspheres Yields of production 14-18

The yields of production microspheres of various batches were calculated using the weight of final product after drying with respect to the initial total weight of the drug and polymer used for preparation of microspheres and percent production yields were calculated as per the formula mentioned below and results are reported in results.

Production yield = Practical mass (microspheres) X 100.....1

Theoretical mass (polymer+drug)

Actual drug encapsulation content and efficiencv14-18

Wherein the calcium chloride solution in which the microspheres were prepared was estimated for its drug content through UV spectroscopy by taking its absorbance at 247nm and the amount of unloaded drug was estimated, then determined amount of drug was deducted from the total quantity of drug added initially to obtain the amount of drug which is encapsulated. Encapsulation efficiency was determined by direct method wherein the microspheres were immersed in the water for 24 hours with constant shaking which would result in the extraction of drug from the microspheres in water, which is then quantitatively estimated trough UV spectroscopy by taking its absorbance at 247nm and the value thus obtained is used to determine the encapsulation efficiency of the microspheres using the formula mentioned below and encapsulation efficiency values were reported in results.

Percent encapsulation efficiency = Actual drug content(mg) X 100.....2 Total

microspheres

Morphology of microspheres14-18

The shape and size of microspheres of the optimized batches was determined through optical microscope and through SEM (cameca, france model-SV30). Results are reported results.

Swelling studies11-14

The swelling ability of the uncoated microspheres

in physiological media was determined by immersing an accurately weighed amount (500 mg) of microspheres in a little excess of 100 ml of phosphate buffer (pH 6.8) and kept for 24 h. equation was to compute the degree of swelling. where Ssw = percent swelling of microspheres. Wo = initial weight of microspheres, Ws = weight of microsphere after swelling.

In vitro release study19-24

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In vitro drug release was studied by dissolution method using dissolution apparatus I (basket). The dissolution was performed in 900 mL (v) acidic buffer pH 1.2. the temperature was maintained at 37 ± 0.5 °C and the speed of basket was kept at 100 rpm during dissolution study. Microspheres filled in capsule and placed in dissolution medium. At appropriate time intervals, 5 mL of the solution was withdrawn, filtered, and the absorbance of samples was measured on UV spectrophotometer (Jasco V-630,Japan) at 224 nm, while an equal volume of fresh dissolution medium was added into the apparatus. Dissolution tests were performed in triplicate. The % drug release was calculated by PCP disso software and reported in results.

Study of release mechanism by Curve fitting

Release data were fitted to various mathematical models for describing the release mechanism from buccal tablets; Korsmeyer-Peppas (Eq. (4)), zeroorder (Eq. (5)) and Higuchi release models (Eq. (6)). And reported in results.

Mt/M∞	=	kKPtn
	4	
Where, Mt/M∞	= fraction of dr	ug released at
time't';		
kKP = release rate con	stant;	
n = releas	se exponent.	
Mt = M0 + k0t	-	
5		
Where, $Mt = A$	Amount of dru	g released at
time't';		-
M0 = co	ncentration of	drug in the
solution at t=0;		C
k0 = zero	o-order release	constant.

t1/2 Mt = kН

Where, Mt = Amount of drug release at time



of

mass

'√t';

kH = Higuchi release constant.

All curve fitting, simulation and plotting was carried out by using disso software (PCP V3). The mechanism of the drug release is discussed in results.

In vitro mucoadhesion strength determination of microparticles 25-29

A freshly excised sheep's stomach was used. Prior to the study, the mucus surface of the tissue was rinsed with normal saline. The tissue was pinned unto a polyethene support inclined at an angle of 600. A beaker was placed directly under the base of the polyethene to collect the micropaticles as they got detached from the tissue. A 100 mg quantity of the microparticles formulated with various ratios of the polymers was placed on the trough of the mucus surface of the tissue and allowed to hydrate for 15 min for microparticle-mucin interaction to take place. A 100 ml volume of SGF was allowed to flow over the tissue at the rate of 40 drops/min. The weight of the microparticle detached (washed out) calculated as a percentage of the original weight was used as a measure of mucoadhesion. And results are reported.

In vivo studies30-34

1. Weight count method

In this technique 5 groups of four number of Albino rats fasted overnight & 100mg of microspheres suspension administer to these rats through needle, then these rats sacrifies at an interval of 0, 4, 8, 12 h respectively. Then after dissection their stomach region isolate & cut open longitudinally to note the weight of microspheres adhering to the stomach & intestine region, which give their adhesive strength using the formula given below.

Were, No = Weight of microspheres hydrated with little amount of water

Ns = Weight of microspheres detaching from mucosal surface.

And results are reported.

		Formulations					
		F1	F2	F3	F4	F5	
*	1	24.119 ±0.83	29.556 ±1.62	24.212 ±1.06	20.81±0.39	26.833 ±0.39	
Perce	2	27.921 ±0.52	41.863 ±1.52	28.033 ±0.41	23.91±0.34	30.446 ±0.34	
nt	3	32.521 ±1.37	56.034 ±0.46	33.022 ±0.25	32.54±0.33	38.293 ±0.33	
drug	4	38.257 ±0.41	60.909 ±0.20	38.419 ±0.17	38.40±0.17	45.876 ±0.17	
releas	5	45.655 ±0.65	66.800 ±0.38	46.391 ±0.24	46.81±0.45	55.492 ±0.45	
e	6	53.740 ±0.79	71.621 ±0.54	53.962 ±0.92	60.51±0.31	61.955 ±0.79	
	7	64.740 ±1.49	77.383 ±1.05	64.778 ±1.03	76.21±1.20	67.165 ±1.49	
	8	68.594 ±1.02	83.267 ±0.89	68.706 ±0.35	85.82±0.32	72.633 ±1.02	
	9	76.941 ±0.99	85.950 ±1.27	77.017 ±1.06	93.72±0.29	81.002 ±0.99	
	10	83.795 ±0.41	89.145 ±0.45	83.732 ±0.39	94.30±1.08	94.202 ±0.41	
	11	88.924 ±0.18	95.115 ±0.40	95.005 ±0.76	94.82±1.21	94.619 ±0.18	
	12	96.006 ±0.35	95.845 ±1.64	95.073 ±1.02	94.88±0.92	94.849 ±0.54	
Producty yield (%	tion 6)	84.01	82.80	78.40	82.60	81.00	
Actual	drug	77.22	76.19	79.89	80.14	79.63	
content	- (%)						
Encaps efficien	ulation cy (%)	76.89	77.49	78.13	82.10	77.93	

Factorial batches dissolution studies for ionic gelation technique

*Represents mean ± S.D. (n = 3)



		Formulations						
	Time (hr)	F6	F7	F8	F9			
*Percent drug	1	25.93±0.34	26.833 ±0.56	26.833 ±0.22	25.574 ±0.11			
release	2	37.00±0.31	37.013 ±0.80	30.446 ±0.23	33.208 ±0.38			
	3	46.71±0.34	52.213 ±1.04	38.293 ±0.29	45.423 ±0.28			
	4	66.36±0.42	66.403 ±0.29	45.876 ±0.12	53.834 ±0.18			
	5	71.18±0.08	76.012 ±0.23	55.492 ±0.22	61.993 ±0.19			
	6	76.95±0.54	85.607 ±0.17	61.955 ±0.56	74.354 ±0.50			
	7	82.83±0.31	94.410 ±0.90	93.107 ±0.29	82.11 ±0.54			
	8	85.51±1.64	93.989 ±0.57	93.626 ±1.07	93.775 ±1.29			
	9	88.71±0.59	94.498 ±0.62	94.135 ±0.67	94.284 ±1.40			
	10	91.86±0.87	95.006 ±0.64	94.645 ±0.63	94.812 ±1.23			
	11	95.39±0.59	95.423 ±0.61	94.951 ±1.17	95.349 ±0.74			
	12	95.90±0.89	95.838 ±0.54	94.959 ±0.41	95.858 ±0.45			
Production yield	(%)	79.41	80.71 82.70		81.72			
Actual drug content(%)		77.44	81.22 78.51		81.59			
Encapsulation (%)	efficiency	78.86	82.25	77.29	79.13			

Table6: Data of release study flurbiprofen from factorial batches

Discussion

In vitro dissolution study of the microspheres indicates thatFormulation f1 is combination of 100:100 HPMC K4M and sodium alginate shows 100% release upto 12 h. f2 is combination of 100:80 HPMC K4M and sodium alginate shows 100% release upto 11 h. f3 is combination of 100:60 HPMC K4M and sodium alginate shows 100% release upto 11 h.

Formulation f4 is combination of 80:100 HPMC K4M and sodium alginate shows 100% release upto 11 h f5 is combination of 80:80 HPMC K4M and sodium alginate shows 100% release upto 10 h f6 is combination of 80:60 HPMC K4M and sodium alginate shows 100% release upto 9 h.

Formulation f7 is combination of 60:100 HPMC K4M and sodium alginate shows 100% release upto 7h f8 is combination of 60:80 HPMC K4M and sodium alginate shows 100% release upto 7h f9 is combination of 60:60 HPMC K4M and sodium alginate shows 100% release upto 7h. From above

discussion it was clear that the as we increases the concentration of polymer HPMC K4M release of drug was retarded.

From above discussion formulation f1 was the optimized formulation.

Yield of production, Actual drug content and encapsulation efficiency

The production yields of microspheres prepared through the ionic gelation technique is found in the range of 78-84%. Actual drug content and encapsulation efficiency or drug entrapment efficiency of the microspheres prepared by ionic gelation technique was found to be 75-85%. It is not up to 100% because in ionic gelation technique microspheres prepared in external aqueous solution of calcium chloride and since drug is water soluble, most of the drug gets diffused in this aqueous solution.









B.



Figure1: Dissolution profile of A. F1-F3 B. F4-F6 C. F7-F9 formulations for factorial batches

Release mechanism

In order to understand the complex mechanism of drug release from mucoadhesive microspheres, the in vitro flurbiprofen release data fitted to korsmeyer-peppas release model. For non fickian release, the values of n falls between 0.5 to 1; while in case of fickian diffusion, n=0.5; For zero order release (case II transport), n=1; and for super case

II transport, n>1(15). The values of n were between 0.5-1 (Table III), indicating that the release of flurbiprofen was found to be non-fickian diffusion. These formulations follows korsmeyer-peppas, zero order, and higuchi model. The table no. shows the R2 (correlation coefficient) and constants for the korsmeyer-peppas, zero order, and higuchi models.



	1	, , , , , , , , , , , , , , , , , , ,		, , , , , , , , , , , , , , , , , , ,		1
Formulation	Korsmeyer-		Zero order		Higuchi	
	Peppas					
	*Kkp (h-n)	*R2	*K0 (h1)	*R2	*KH (h1/2)	*R2
F1	19.2353 ±	0.9717	8.5290 ±	0.9729	20.5343	0.9734
	1.065	±0.001	0.335	±0.002	±0.687	±0.002
F2	30.9127 ±	0.9958	9.7815	0.8041	27.4527	0.9823
	2.600	±0.003	±0.319	±0.003	±0.784	±0.008
F3	19.4774 ±	0.9735	8.5003	0.9697	23.6584	0.9788
	1.104	±0.001	±0.244	±0.006	±0.538	±0.007
F4	27.0174 ±	0.9854	10.0069	0.8348	29.0911	0.9864
	0.969	±0.002	±0.172	±0.011	±0.895	±0.001
F5	22.5523 ±	0.9824	9.1331 ±	0.9467	22.4401	0.9763
	2.315	±0.010	0.145	±0.010	±0.623	±0.013
F6	16.5141 ±	0.9732	9.3929 ±	0.9664	20.2178	0.9725
	0.256	±0.009	0.348	±0.008	±0.879	±0.014
F7	28.3458 ±	0.9734	10.5075	0.7907	18.3309	0.9467
	0.775	±0.002	±0.162	±0.002	±0.997	±0.011
F8	22.2536	0.9628	9.8672 ±	0.9152	18.8960	0.9489
	±3.363	±0.001	0.210	±0.001	±0.994	±0.009
F9	24.2871 ±	0.9825	10.1515	0.8842	19.0023	0.9633
	0.969	±0.002	±0.172	±0.002	±0.978	±0.006

 Table7: Data for study of release mechanism by curve fitting analysis

*Represents mean ± S.D. (n = 3)

Factorial batches dissolution studies spray dry

Table8: Data of release study flurbiprofen from factorial batches

		Fo	Formulations						
		F1	F2	F3	F4	F5			
* Percent	1	24.753 ±0.21	26.674 ±0.21	29.548 ±1.45	21.81±0.39	27.813 ±0.54			
drug	2	31.346 ±0.20	33.218 ±0.35	41.863 ±1.54	24.91±0.34	30.546 ±0.34			
release	3	39.293 ±0.28	46.423 ±0.28	57.134 ±0.46	32.74±0.33	39.293 ±0.33			
	4	46.876 ±0.12	54.834 ±0.18	61.909 ±0.20	39.40±0.17	45.886 ±0.17			
	5	55.592 ±0.26	61.853 ±0.19	66.800 ±0.38	45.81±0.45	56.492 ±0.45			
	6	62.555 ±0.65	75.354 ±0.55	71.621 ±0.54	61.51±0.31	63.955 ±0.79			
	7	93.121 ±0.29	83.11 ±0.54	78.383 ±1.05	77.21±1.20	67.765 ±1.49			
	8	93.726 ±1.07	93.982 ±1.28	83.467 ±0.89	85.72±0.32	72.633 ±1.02			
	9	94.035 ±0.67	94.184 ±1.40	95.255 ±0.44	93.82±0.29	82.102 ±0.99			
	10	94.545 ±0.66	94.742 ±1.23	95.310 ±0.32	94.20±1.08	94.401 ±0.42			
	11	94.931 ±1.17	94.949 ±0.74	95.422 ±0.40	94.72±1.21	94.719 ±0.18			
	12	94.960 ±0.43	95.558 ±0.55	95.556 ±1.64	94.78±0.82	94.849 ±0.14			
Production yield (%)		31.55	35.25	44.25	33.74	42.98			
Encapsulation efficiency (%)		60.24	71.44	80.11	72.14	79.55			
Swelling index (%)		204±8	212±6	260±5	158±4	169±4			

*Represents mean ± S.D. (n = 3)



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		Formulations					
		F6	F7	F8	F9		
*Percent drug	1	25.33±0.31	26.003 ±0.14	24.512 ±1.16	24.619 ±0.45		
release	2	36.00±0.32	37.253 ±0.80	29.013 ±0.41	27.721 ±0.52		
	3	46.81±0.34	53.213 ±1.04	33.332 ±0.25	32.561 ±1.37		
	4	67.36±0.42	65.403 ±0.29	38.429 ±0.17	38.557 ±0.41		
	5	71.18±0.08	76.212 ±0.23	47.391 ±0.24	46.655 ±0.65		
	6	76.85±0.51	84.624 ±0.17	53.882 ±0.92	53.780 ±0.79		
	7	81.83±0.31	94.110 ±0.77	65.778 ±1.23	65.721 ±1.49		
	8	85.11±1.64	94.252 ±0.27	69.706 ±0.35	69.594 ±1.02		
	9	86.71±0.59	94.457 ±0.62	77.517 ±1.06	76.741 ±0.99		
	10	95.59±0.59	95.106 ±0.44	83.532 ±0.39	83.195 ±0.41		
	11	95.66±0.54	95.553 ±0.61	95.273 ±1.87	88.544 ±0.24		
	12	95.80±0.19	95.778 ±0.44	95.410 ±1.51	96.146 ±0.45		
Production yield (%)	51.65	39.54	55.64	62.95		
Encapsulation e (%)	fficiency	88.12	73.25	80.54	92.57		
Swelling index (%)		175±6	118±5	130±4	148±4		

Discussion

In vitro dissolution study of the microspheres indicates thatFormulation f1 is combination of 1:2:0 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 0.0 % shows 100% release upto 7.0 h. f2 is combination of 1:2:1 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 0.0 % shows 100% release upto 8.0 h. f3 is combination of 1:2:2 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 0.0 % shows 100% release upto 9.0 h.

Formulation f4 is combination of 1:2:0 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 20.0 % shows 100% release upto 9.0 h f5 is combination of 1:2:1 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 20.0 % shows 100% release upto 10 h f6 is combination of 1:2:2 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 20.0 % shows 100% release upto 10 h.

Formulation f7 is combination of 1:2:0 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 30.0 % shows 100% release upto 7h f8 is combination of 1:2:1 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 30.0 % shows 100% release upto 11h f9 is combination of 1:2:2 Flurbiprofen : Sodium alginate : Sodium CMC & cross linking agent 30.0 % shows 100% release upto 12h. From above discussion it was clear that the as we increases the concentration of polymer & cross linking agent release of drug was retarded.

From above discussion formulation f9 was the optimized formulations.







Figure2: Dissolution profile of A. F1-F3 B. F4-F6 C. F7-F9 formulations for factorial batches

Yield of production, Actual drug content and encapsulation efficiency

The production yields of microspheres prepared through the spray dry technique is found in the

In vitro mucoadhesive strength determination

range of 30-62%. Actual drug content and encapsulation efficiency or drug entrapment efficiency of the microspheres prepared by spray dry technique was found to be 60-92%.

Table9: In vitro data for mucoadnesive strength determination									
SR. NO	WEIGHT	' (mg) (OF MICRO	SPHERES	%				
	REMAINING ON GASTRIC MUCOSA			MUCOADHESIVE					
			STRENGTH						
	3h	6h	9h	12h					
F1 (Ionic	45	41	37	34	78.50				
gelation)									
F9 (Spray dry)	44	41	35	32	76.00				



Ex vivo mucoadhesive strength determination

Tuble 10. In vivo data for indebudicesive strength determination					
Sr.no	Weight (mg) of microspheres remaining on the rat				%
	stomach				mucoadhesive
	Time (h)				strength
Optimized	0	4	8	12	
Ionic gelation	98.30	81.31	72.31	67.44	79.84
Spray dry	98.12	80.34	74.21	64.32	78.24

Table10: In v	ivo data for muco	adhesive strengtl	determination
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From both in vitro and in vivo mucoadhesive strength determination tests it was cleared that in ionic gelation & Spray dry technique optimized formulations shows 79.84% & 78.245 mucoadhesive strength respectively. Ionic gelation formulation comparising of 100:60 of HPMC K4M:sodium alginate & Spray dry formulation comparising of 1:2:2 ratio of flurbiprofen: Sodium alginate: Sodium CMC it retard the release of drug

up to 12 hours due to high mucoadhesive strength.

Morphology of microspheres

Morphological study of microspheres done using SEM & microspheres was studied which shows shape of microspheres almost spherical shown in fig no.3 and size shown in table no.9





Fig3: Morphology of microspheres A. Ionotropic gelation B. Spray dry

Table 11: Size of optimized inici ospheres				
FORMULATIONS	SIZE in µm	SHAPE		
SIZE in µm(Gelation)	55.32-67.12	Almost spherical		
SIZE in µm(Spray Dry)	11.32-12.50	Almost spherical		

Table11: Size of optimized microspheres

Optimization of mucoadhesive microspheres formulations(Ionic Gelation Method)

Effect of formulation variables.

Effect of formulation variables on T50%

The model terms for response Y1 (T50%) were found to be significant with F value of 4.73 (p<0.0047). In this case all the factors were found to be significant and the model describing T50% can be written as;

Y1 = 2.96 + 0.53X1-0.29 X2+ 0.27 X1 X2 + 0.46 X12 + 1.10 X22

As the amount of X1 and X2 increases the

corresponding T50% (time required to release 50% of the drug) also increases The Fig 10 shows the response surface plot. It indicates at all the high levels of X1 and X2 the T50% value is high, As discussed above this behavior is due to increase in amount of sodium alginate and HPMC K4M forms a high viscous gel matrix and thus decreases the drug release and hence T50% value increases, while HPMC K4M forms pores in the formed matrix and will increases the drug release thus decreases the T50% value. The Fig 11 shows the graph of predicted verses actual data.







Figure 11 Linear correlation plots between actual and predicted values for T_{arra} (Y1)

Effect of formulation variables on T90%

The model terms for response Y2 (T90%) were found to be significant with F value of 10.06 (p<0.0001). In this case all the factors were found to be significant and the model describing T50% can be written as;

Y2 = -5.79 + 0.68X1 - 14.83X2 + 0.99 X1X2 + 15.32 X12 + 16.12 X22

As the amount of X1 and X2 increases the corresponding T90% (time required to release 90% of the drug) also increases The Fig 12 shows

the response surface plot. It indicates at all the high levels of X1 and X2 the T50% value is high, As discussed above this behavior is due to increase in amount of sodium alginate and HPMC K4M forms a high viscous gel matrix and thus decreases the drug release and hence T50% value increases, while HPMC K4M forms pores in the formed matrix and will increases the drug release thus decreases the T90% value. The Fig 13 shows the graph of predicted verses actual data.



Figure 12: Response surface plot showing effect of formulation variables on T90%



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Figure 13: Linear correlation plots between actual and predicted values for T90% (Y2)

Effect of formulation variables on the drug release at 8 hr. (Y3)

The quadratic model was found to be significant with an F value 27.44 (P<0.0001). In this case X1, X2 was found to be significant and the model describes the percent flurbiprofen release at 8h can be written as;

Y3 = 82.91 - 0.30X1 + 10.17 X2

As the concentration of mucoadhesive polymer (sodium alginate and HPMC K4M) increased it causes an increase in viscosity of swollen gel matrix, which contributes more hindrance for drug diffusion and thus decreases the release rate. The combined effect of X1, X2 shown in response surface plot (Fig 14) In this plots it was observed that the increasing amount of sodium alginate and HPMC K4M causes the decreases in the drug release, due to formation of high viscous gel matrix. Thus the factors X1 and X2 have negative effect on the drug release. The Fig 15 Shows a graph of observed verses predicted values. The sodium alginate and HPMC K4M have negative effect on drug release, due to increased viscosity and gel strength. The swelling of sodium alginate may be due to uncharged –COOH group which forms hydrogen bonds with imbibing water and also holds water inside the gel matrix. Increasing amount of HPMC K4M which contains –OH groups will may increases the formation of hydrogen bonding and form a gel matrix network with sodium alginate.



Figure 14: Response surface plot showing effect of formulation variables on percent drug release at 8 h

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Optimization of mucoadhesive microspheres formulations(Spray Dry Method)

Effect of formulation variables.

Effect of formulation variables on T50%

The model terms for response Y1 (T50%) were found to be significant with F value of 4.88 (p<0.0048). In this case all the factors were found to be significant and the model describing T50% can be written as;

Y1 = 2.87 + 0.51X1-0.28 X2+0.26 X1 X2 + 0.48 X12 + 1.14 X22

As the amount of X1 and X2 increases the corresponding T50% (time required to release

50% of the drug) also increases The Fig 3 shows the response surface plot. It indicates at all the high levels of X1 and X2 the T50% value is high, As discussed above this behavior is due to increase in amount of polymers (sodium alginate and Sodium CMC) & cross linking agent forms a high viscous gel matrix and thus decreases the drug release and hence T50% value increases, while Sodium CMC forms pores in the formed matrix and will increases the drug release thus decreases the T50% value. The Fig 4 shows the graph of predicted verses actual data.



Figure 3 Basponse surface plot showing effect of formulation variables on $T_{\rm atta}$

Effect of formulation variables on T90%

The model terms for response Y2 (T90%) were found to be significant with F value of 10.11 (p<0.0001). In this case all the factors were found to be significant and the model describing T50%



Figure 41 lasar correlation plats between actual and predicted values for T $_{\rm pro}(\rm VJ)$

can be written as;

Y2 = -5.79 + 0.68X1 - 14.83X2 + 0.99 X1X2 + 15.32 X12 + 16.12 X22 As the amount of X1 and X2 increases the

As the amount of X1 and X2 increases the corresponding T90% (time required to release



90% of the drug) also increases The Fig 5 shows the response surface plot. It indicates at all the high levels of X1 and X2 the T50% value is high, As discussed above this behavior is due to increase in amount of polymers (sodium alginate and Sodium CMC) & cross linking agent forms a high viscous gel matrix and thus decreases the drug release and hence T50% value increases, while Sodium CMC forms pores in the formed matrix and will increases the drug release thus decreases the T90% value. The Fig 6 shows the graph of predicted verses actual data.



Effect of formulation variables on the drug release at 8 hr. (Y3)

The quadratic model was found to be significant with an F value 28.22 (P<0.0001). In this case X1, X2 was found to be significant and the model describes the percent flurbiprofen release at 8h can be written as;

Y3 = 81.76 - 0.29X1 + 11.45 X2

As the concentration of mucoadhesive polymer (sodium alginate and Sodium CMC) increased it causes an increase in viscosity of swollen gel matrix, which contributes more hindrance for drug diffusion and thus decreases the release rate. The combined effect of X1, X2 shown in response surface plot (Fig 7) In this plots it was observed that the increasing amount of Sodium CMC causes the decreases in the drug release, due to formation of high viscous gel matrix. Thus the factors X1 and X2 have negative effect on the drug release. The Fig 8 Shows a graph of observed verses predicted values. The sodium alginate and Sodium CMC have negative effect on drug release, due to increased viscosity and gel strength. The swelling of sodium alginate may be due to uncharged –COOH group which forms hydrogen bonds with imbibing water and also holds water inside the gel matrix. Increasing amount of Sodium CMC which form a gel matrix network with sodium alginate.



ANOVA, Pure error, Lack of fit

The results of ANOVA in Table 14 for the dependent



variables demonstrate that the model was significant for all response variables. Regression analysis was carried out to obtain the regression coefficient (Table 15) and effects as follows; all factors found to be significant for response Y1, similarly only X1, X2 and X1X2 were found for Y2, the X1, X2 were found significant for Y3. The above results conveyed us that the amount of sodium alginate, HPMC K4M plays important role in formulation of mucoadhesive microspheres of flurbiprofen. Thus appropriate range of these ed mucoadhesive microspheres with good bioadhesive strength and drug release. The data of pure error and lack of fit are summarized in Table 14 The residuals are the difference in the observed and predicted value. Since computed F values were respectively less than critical F values, denotes nonsignificance of lack of fit.

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Source	d.f.	Sum	Mean	F value	Probability
		square	square		
T50% (h)					
Model	5	15.89	3.18	4.73	0.0047
Residual	21	14.10	0.67		
Total	26	30.00			
Lack of fit	3	13.82	4.61	295.79	< 0.0001
Pure error	18	0.28	0.016		
T90% (h)					
Model	5	6948.06	1389.61	10.06	< 0.0001 *
Residual	21	2901.00	138.14		
Total	26	9849.06			
Lack of fit	3	2900.00	966.67	17347.34	< 0.0001
Pure error	18	1.00	0.056		
NF release at 8 h (%)					
Model	2	1863.81	931.91	27.44	< 0.0001 *
Residual	24	815.18	33.97		
Total	26	2678.99			
Lack of fit	6	804.28	134.05	221.14	< 0.0001
Pure error	18	10.90	0.61		
	Source T50% (h) Model Residual Total Lack of fit Pure error T90% (h) Model Residual Total Lack of fit Pure error NF release at 8 h (%) Model Residual Total Lack of fit Pure error	Sourced.f.T50% (h)5Model5Residual21Total26Lack of fit3Pure error18T90% (h)5Residual21Total26Lack of fit3Pure error18T90% (h)21Model5Residual21Total26Lack of fit3Pure error18NF release at 8 h (%)18Model24Total26Lack of fit6Pure error18	Source d.f. Sum square T50% (h) 5 15.89 Model 5 15.89 Residual 21 14.10 Total 26 30.00 Lack of fit 3 13.82 Pure error 18 0.28 T90% (h) - - Model 5 6948.06 Residual 21 2901.00 Total 26 9849.06 Lack of fit 3 2900.00 Pure error 18 1.00 NF release at 8 h (%) - - Model 2 1863.81 Residual 24 815.18 Total 26 2678.99 Lack of fit 26 2678.99 Lack of fit 6 804.28 Pure error 18 10.90	Source d.f. Sum square Mean square T50% (h) - - - Model 5 15.89 3.18 Residual 21 14.10 0.67 Total 26 30.00 Lack of fit 3 13.82 4.61 Pure error 18 0.28 0.016 T90% (h) - - - Model 5 6948.06 1389.61 Residual 21 2901.00 138.14 Total 26 9849.06 Lack of fit 3 2900.00 966.67 Pure error 18 1.00 0.056 NF release at 8 h (%) - - - Model 2 1863.81 931.91 Residual 24 815.18 33.97 Total 26 2678.99 Lack of fit 6 804.28 134.05 Pure error 18 10.90 0.61	Source d.f. Sum square Mean square F value T50% (h) 5 15.89 3.18 4.73 Model 5 15.89 3.18 4.73 Residual 21 14.10 0.67 Total 26 30.00 Lack of fit 3 13.82 4.61 295.79 Pure error 18 0.28 0.016 T90% (h) Model 5 6948.06 1389.61 10.06 Residual 21 2901.00 138.14 Total 26 9849.06 Lack of fit 3 2900.00 966.67 17347.34 Pure error 18 1.00 0.056 NF release at 8 h (%) Model 2 1863.81 931.91 27.44 Residual 24 815.18 33.97 Lack of fit 6

 Table11: Data of ANOVA study for dependent variables from 32 factorial design



Table 12: Data of ANOVA study for results in analyzing lack of fit and pure

Optimization

A numerical optimization technique by the desirability approach was used to generate the optimum settings for formulation. The process was optimized for dependent variables Y1-Y4. The optimized formula arrived by targeting the Y1 was targeted at 6 h, Y2 was targeted at 10 h,Y3 was kept at range 70-80% drug release. The optimized

results obtained to give 7 results out of that one formula is shown in Table 16. The results of optimized formula were compared with the predicted values and it was shown in Table 17 which showed good relationship between experimented and predicted values, which confirms the practicability and validity of the model.

Table13: Composition of optimized formulation

Ingredients	Quantities (mg)
Drug	50
Sodium alginate	100
HPMC K4M	60

Table 14: Comparison between the experimented and predicted values for most probable optimal2275formulation

Dependent variables	Optimized formulation			
	*Experimented value	Predicted value		
Sodium alginate	98.908 ± 2.48	98.225		
НРМС К4М	57.23 ± 0.11	57.3833		

Source d.f. Sum Mean square F value Probability square T50% (h) 1 5.08 5.08 7.56 0.0120 X1 1 1.48 1.48 2.20 0.1526 1 0.87 0.2677 X2 0.87 1.30 X1X2 T90% (h) 1 8.28 8.28 0.060 < 0.0001 3959.58 3959.58 < 0.0001 X1 1 28.66 0.0009 X2 1 11.80 11.80 0.085 X1X2 NF release at 8 h (%) 1 1.60 1.60 0.047 0.8298 X1 1 1862.21 1862.21 54.83 < 0.0001 X2

*Represents mean ± S.D. (n = 3)

Summery & Conclusion

The results so far obtained during this investigation encouraged us to derive the following conclusions The production yield of microspheres prepared by ionic gelation technique was found in the range of 78-84% which is reliable.

The production yield of microspheres prepared by spry drying technique was found in the range of 30-62 % which is reliable



The encapsulation efficiency of microspheres prepared by ionic gelation technique was found in the range of 75-85% it is not 100% because during preparation of microspheres some drug lost in external media.

The encapsulation efficiency of microspheres prepared by spry drying technique was found in the range of 60-92% it is not 100% because during preparation of microspheres some drug lost in external media.

The in vitro release profile of Flurbiprofen from optimized formulations in ionic gelation technique were F1 shows retardation of release up to 12 hours shows good controlled release.

The in vitro release profile of Flurbiprofen from optimized formulations in spray drying technique were F9 shows retardation of release up to 12 hours shows good controlled release.

The in vitro Flurbiprofen release data fitted to korsmeyer-peppas release model also shows zero order and higuchi model.

The in vitro mucoadhesive strength of optimized formulations of ionic gelation technique were for F1 78.50% & F3 76.50% which shows good mucoadhesion.

The ex vivo mucoadhesive strength of optimized formulations of ionic gelation & spray drying technique were for F1 79.84% & F9 78.24% which shows good mucoadhesion.

The size of microspheres prepared by ionic gelation & spray drying technique was found for F1 is $55.32-67.12 \ \mu m \&$ for F9 $11.32-12.50 \ \mu m$

References

- Gelbert S. Banker, Christopher T. Rhodes.1996, Modern pharmaceutics, New York Marcel Dekker , fourth edition,:. 501-516.
- James C.B., James S., Yie W.C., 2005, Encyclopedia of pharmaceutical technology, 2nd Edition, vol.-1 sengshang lin., New York,: 810.
- Stanley S.Davis.2005, Formulation strategies for absorbtion window, Drug Discovery Today-vol 10(4),: 34-55
- Moes A. J. 2003, Gastric Retention Systems for Oral Drug Delivery, Business Brifefing, Pharmatech,:157-159
- Chowdhary K P R and Rao Y S. 2004, Mucoadhesive microspheres for controlled drug delivery, Biol. Pharm. Bull. 27(11),: 1717-1724

Brahamankar D M, Jaiswal S B.2005, Biopharmaceutics & Pharmacotherapeutics, Vallabh Prakashan,:. 335-339

- Jain N K.2008, Advances in controlled and noval drug delivery, CBS publication,:.1-10, 13-18, 19
- Mathiowitz E, Chickering, D. E, D.E., 1992, Definition, Mechanisms and Theories Of Bioadhesion, in; E. Mathiowitz,D.E. Chickering, C.M. lehr(eds), Bioadhesive drug delivery system :fundamentals, novel approaches and development, Marcel Dekker, New York,: 1-10,16-19
- Das M. K , Senapati. P. C. 2004, Furosemide Loaded

AlginateMicrosphere prepared by Ionic Cross linking Technique Morphology & Release Characteristics,: 122-125

- Yeong Woo Jo, Byung Ho Woo, Azar M. Hazrati, and Patrick P. DeLuca.2005, Use of Pharmasep Unit for Processing Microspheres, AAPS PharmSciTech; 2 (1),: 11-23
- Avachat A, Ahire V. J. 2001, Charaterization& Evaluation Of Spray Dried Co-Processed Excipients & Their Application in Solid Dosage Forms., International Journal of Pharmaceutics, : 153–161
- Bhalerao SS, Lalla JK, Rane MS.2001, Study of processing parameters influencing properties of diltiazem hydrochloride. J Microencap :18,: 299-307
- Dani BA, DeLuca PP.2001, Preparation, Characterization, and In Vivo Evaluation of Salmon Calcitonin Microspheres. AAPS PharmSciTech. ; 2(4): article 22,: 54-64
- Jun Kunisawaa, Akiko Okudairaa, Yasuo Tsutusmia, Ichiro Takahashib,TsuyoshiNakanishia, Hiroshi Kiyonob, Tadanori Mayumia. 2001, Characterization of mucoadhesive microspheres for the induction of mucosal and systemic immune responses, Vaccine 19,: 23-41,45-56
- Bredenberg S, Duberg M, Lennernas B, Lennernas H, Pettersson A, Westerberg M, Nystrom C, 2003, Eur J Pharm Sci, 20:327-334
- Mani Prabakaran, Shaoqin Gong.2003, Novel thiolated carboxymethyl chitosan-g-b-cyclodextrin as mucoadhesive hydrophobic drug delivery carriers, Carbohydrate Polymers 73,: 117–125
- Attama A A, Nwabunze J.2007, Mucuna gum microspheres for oral delivery of glibenclamide: In vitro evaluation, . Acta Pharm. 57,: 161
- Khar, R, Ahuja, A, 1997. Mucoadhesive Drug Delivery System, Drug Dev. Ind. Pharm., 23 (5),: 489-515.
- Swarbrik, J, Boylon J. C., 2002, In K.R., Kamath, K., Park, Encyclopedia of pharmaceutical Technology. 2nd Ed.; INC: New York, Marcel Dekker,: 133.
- Khar, R, Ahuja, A, Javed, A, 1997, In N.K., Jain., Mucoadhesive drug delivery in Controlled and Novel Drug Delivery, 3rd ed., CBS publishers and distributors, New Delhi,:23-26
- Edith, M, Donald, E.C, Claus, M.L,1999, Bioadhesive drug delivery systems, 3rd edition, New York, Marcel Dekker,:. 89-94.
- Myung-Kwan Chun, HongkeeSah ,Hoo-Kyun Choi. 2006, Preparation of mucoadhesive microspheres containing antimicrobial agents for eradication of H. pylori, International Journal of Pharmaceutics 297,: 172–179
- Jianjun Cheng,1 Benjamin A. Teply, Seok Yoon Jeong, Christopher H.2006, Yim, Dennis Ho, Ines Sherifi, Sangyong Jon, Omid C. Farokhzad, Ali Khademhosseini, and Robert S. Langer. Magnetically Responsive Polymeric Microparticles for Oral Delivery of Protein Drugs, Pharmaceutical Research, Vol. 23, No. 3: 557-564.
- Mu, L., Teo, M. M., Ning, H.Z., Tan, C.S. and Feng, S.S., J. Control. Release, 2005, 103, 565
- Chein, Y.W., 1992, Novel drug delivery systems, 2nd edition, New York, Marcel Dekkar. :225-228
- Montisci M.-J, Giovannuci G., Duche^{ne} D., Ponchel G.2001, Covalent coupling of asparagus pea and tomato lectins to poly(lactide) microspheres, International Journal of Pharmaceutics 215,: 153–161
- Sanju Dhawan, Anil Kumar Singla, and Vivek Ranjan Sinha.2004, Evaluation of Mucoadhesive Properties of Chitosan Microspheres Prepared by Different Methods,



AAPS PharmSciTech ; 5 (4) Article 67,: 13-23

- Lian-Yan Wang, Yong-Hong Gu, Qing-Zhu Zhou, Guang-Hui Ma,Yin-Hua Wan , Zhi-Guo Su.2006, Preparation and characterization of uniform-sized chitosan microspheres containing insulin by membrane emulsification and a two-step solidification process, Colloids and Surfaces B: Biointerfaces 50,: 126–135
- Yamada T, Onishi H, and Machida Y, 2001, In Vitro and in Vivo Evaluation of Sustained Release Chitosan Coated Ketoprofen Microparticles, YakugakuZasshi 121(3),: 239–245
- Chowdary K.P.R. and Srinivasa Rao Y.2004, Design and In Vitro and In Vivo Evaluation of Mucoadhesive Microcapsules of Glipizide for Oral Controlled Release: A Technical Note, AAPS PharmSciTech ; 4 (3) Article 39,: 45-51,

Chowdhary K. P. R, Shrinivasa Rao Y.2003, Preparation &

Evaluation of Mucoadhesive Microcapsules of Indomethacin., Saudi Pharmaceutical Journal Vol. II, No.3 ,: 97-103

- Myung-Kwan Chun, Chong-Su Cho, Hoo-Kyun Choi.2005, Mucoadhesive microspheres prepared by interpolymer complexation and solvent diffusion method, International Journal of Pharmaceutics 288,; 295–303
- Jian Wanga, Yasuhiko Tabatab, Dianzhou Bic, Kazuhiro Morimoto.2001, Evaluation of gastric mucoadhesive properties of aminated gelatin microspheres, Journal of Controlled Release 73 ,: 223–231
- Yasunori Miyazaki, Kanako Ogihara, Shigeru Yakou, Tsuneji Nagai, Kozo Takayama.2003, In vitro and in vivo evaluation of mucoadhesive microspheres consisting of dextran derivatives and cellulose acetate butyrate, International Journal of Pharmaceutics 258,: 21–29

