



# Estimation of Diclofenac diethylamine and Plumbagin in a drug delivery system by simultaneous equation method

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## Abstract

**Objective** The aim of this study was to develop and validate UV-Vis spectrophotometric method for simultaneous estimation of Diclofenac diethylamine and Plumbagin in a combined dosage form.

**Method** Solubility studies were performed to select solvent for UV-Vis spectrophotometric analysis.  $\lambda_{max}$  values were selected on the basis of analysis of overlay spectra. Simultaneous equation for estimation of drugs (Diclofenac diethylamine and Plumbagin) in a combined dosage form was developed using value of absorbance and absorptivity.

**Key findings** Methanol was selected as solvent on the basis of solubility study. Two  $\lambda_{max}$  values, 282nm and 418nm were selected as wavelength of detection. Diclofenac diethylamine and Plumbagin both followed Beer's law over concentration ranges of 5 to 25  $\mu\text{g/ml}$  and 5 to 40  $\mu\text{g/ml}$ , respectively.



**Conclusion** The simultaneous equation for estimation of Diclofenac diethylamine and Plumbagin was developed according to ICH guidelines and validated for linearity, accuracy, precision, robustness, limit of detection and limit of quantification.

**Keywords:** UV-Vis Spectroscopy; Simultaneous Equation Method; Diclofenac diethyl amine; Plumbagin

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## 1 Introduction

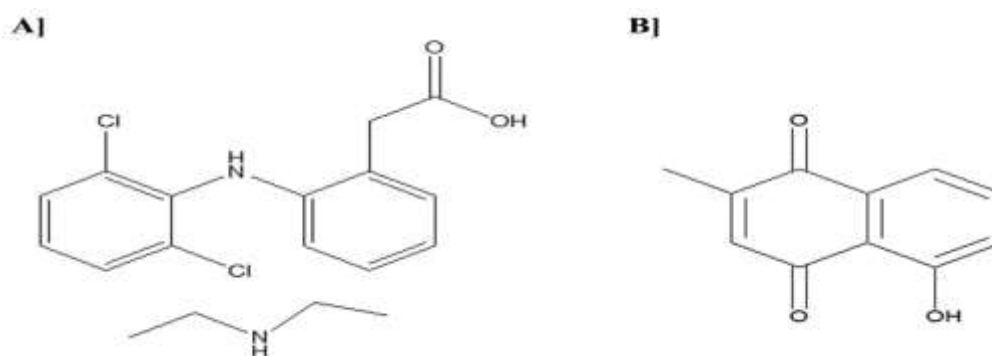
Diclofenac diethylamine (DDEA), 2-[2-(2,6-dichloroanilino) phenyl]acetic acid or  $C_{18}H_{23}Cl_2N_2O_2$  (Fig.1) is an active pharmaceutical ingredient which belongs to the category of non-steroidal anti-inflammatory drugs (NSAID). It is used to treat low back pain, musculoskeletal problems, strain, sprain, arthritis, contusions (bruises), post-traumatic pain, and inflammation associated with these conditions [1]. It acts by inhibiting the activity of an enzyme known as cyclo-oxygenase (COX) that produces the chemical messenger prostaglandin (PG). As a result, less PGs are formed COX blockage, which helps to reduce pain and inflammation [2].

Plumbagin (PLUM), 5-hydroxy-2-methyl-1,4-naphthoquinone or  $C_{11}H_8O_3$  (Fig.1) is a natural

anti-inflammatory compound obtained from roots of *Plumbago zeylanica* [3]. Additionally, in the condition of hepatocellular carcinoma, plumbagin specifically blocks p300-dependent acetylation of histone H3, H4, and p53 while having no impact on PCAF (P300/CBP associating factor) indicating its significance as anti-cancer agent [4].

Literature review reveals that numerous analytical techniques are available for estimation of DDEA and PLUM separately [5-11]. However, not a single method for simultaneous estimation of both compounds has been reported till date. Hence, the objective of this study was to establish a quick, accurate, economical, and validated procedure for quantitative determination of DDEA and PLUM in a combined formulation.

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**Fig 1. Chemical formula arrangement of (A) Diclofenac diethylamine (B) Plumbagin**

## 2 Materials and Methods

### 2.1 Materials

Shimadzu UV-1780 double beam UV-Vis Spectrophotometer was used to analyse the spectral data using UV Probe software. Digital ultrasonic cleaner was used to sonicate the prepared solution. APIs were weighed using digital balance (Citizen, Model CDT 220).

DDEA was provided as a gift sample from Unique Chemicals (Mumbai, India). PLUM was purchased from P.C. Chem (Mumbai, India). Methanol was procured from Loba Chem Pvt. Ltd (Mumbai, India). All other chemicals were of analytical grade.



## 2.2 Method development

### 2.2.1 Selection of solvent

Excess quantity of DDEA and PLUM were dissolved separately in 5ml of each solvent (Water, Acetonitrile, Methanol, Ethanol and Dimethyl sulfoxide) in cyclomixer (CM 101Plus, REMI, India) to determine their solubility. Both drugs showed highest solubility in methanol, therefore selected as solvent for method development.

### 2.2.2 Preparation of standard stock solution

DDEA and PLUM stock solutions, each comprising 200  $\mu\text{g/ml}$  was prepared by

dissolving a fixed quantity (10 mg) of each compound in 50 ml of methanol, followed by 10-minute sonication.

### 2.2.3 Determination of $\lambda_{\text{max}}$ of DDEA and PLUM

The standard solutions of DDEA and PLUM were further diluted to get final concentration of 10  $\mu\text{g/ml}$ . Each solution was scanned independently between 200–700 nm wavelength range against methanol as a blank. DDEA showed  $\lambda_{\text{max}}$  at 282nm and PLUM at 418 nm and 265 nm (Fig. 2).

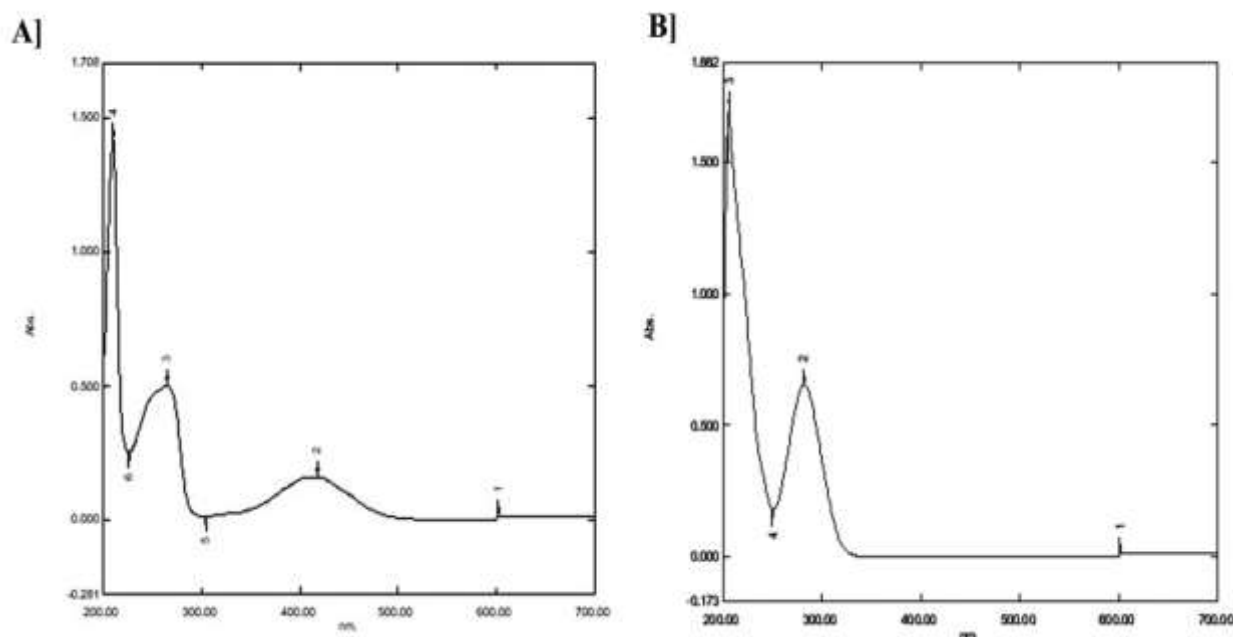
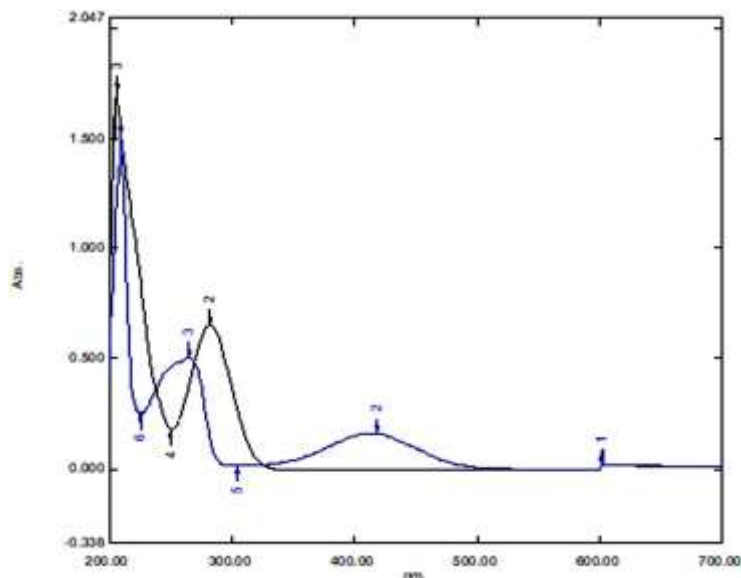


Fig 2. Spectrum of (A) PLUM (B) DDEA

### 2.2.4 Overlay Spectra of PLUM and DDEA

The overlay spectra of DDEA and PLUM was recorded as shown in Fig.3 and two wavelengths 282nm ( $\lambda_{\text{max}}$  of DDEA) and 418nm ( $\lambda_{\text{max}}$  of PLUM) were selected for further investigation.





**Fig 3. Overlay Spectrum of DDEA and PLUM**

### 2.2.5 Development of simultaneous equation

Standard solutions of DDEA (5 - 25 µg/ml) and PLUM (5 - 40 µg/ml) were prepared separately in methanol and their absorbance values were recorded at 282 nm and 418 nm

$$\text{Absorptivity} = \frac{\text{Absorbance}}{\text{Concentration (gm /100 ml)}}$$

Using these absorptivity values simultaneous equation were developed as follows

$$C_x = \frac{A_2 a_1 - A_1 a_2}{a_1 a_2 - a_2 a_1} \dots \dots \dots (1)$$

$$C_y = \frac{A_1 a_2 - A_2 a_1}{a_1 a_2 - a_2 a_1} \dots \dots \dots (2)$$

Where  $C_x$  and  $C_y$  are the concentration of DDEA and PLUM respectively,  $A_1$  and  $A_2$  are the absorbance of sample solution at  $\lambda_{max}$  282 nm and 418 nm respectively,  $a_{x1}$  and  $a_{x2}$  are the absorptivity of DDEA at 282 nm and  $a_{y1}$  and  $a_{y2}$  PLUM at 418 nm [12].

### 2.3 Validation of developed method

The developed method was evaluated for linearity, accuracy, sensitivity and precision in accordance with ICH Q2B guidelines [13].

#### 2.3.1.1 Linearity

Working standard solutions of DDEA and PLUM were prepared in concentration range of 5 to 25 µg /ml and 5 to 40 µg /ml, respectively, and diluted appropriately before analysis. The least square regression approach was used to calculate the linearity from calibration curves.

#### 2.3.1.2 Precision

Repeatability, intraday, and interday variations were adopted to achieve precision. By analysing the concentrations of DDEA and

PLUM at different time intervals under same condition, intra-day precision was estimated. Also the interday precision was determined by analysing the same concentration of solutions on three different days.

#### 2.3.1.3 Accuracy (Recovery studies)

Recovery studies were carried out to examine the accuracy by adding standard drug solutions to pre-analyzed samples at various concentration levels of 80%, 100%, and 120%.

#### 2.3.1.4 Limit of Detection (LOD) and Limit of Quantification (LOQ)

LOD and LOQ of developed method were evaluated after examination of three duplicates of same standard solution.

The LOD may be calculated as



LOD = 3.3 × SD/ Slope

The LOQ may be calculated as

LOQ = 10 × SD/ Slope

Where, SD = Standard Deviation of the intercept

Slope = Slope of the calibration curve

**2.3.1.5 Robustness**

It indicates the ability of a technique to remain undisturbed by minute, intentional changes to the parameters of the method.

**2.4 Assay method of gel formulation**

The total quantity of drug incorporated in gel was 1 % w/w (0.75 gm DDEA + 0.25 gm PLUM). In order to estimate the concentration of DDEA and PLUM in gel formulation, 0.5 g of formulation was dissolved in sufficient quantity of methanol (5ml) and the required volume was made up to 50 ml in a volumetric flask to get the final concentration of 0.075

mg/ml and 0.025 mg/ml respectively. Further, the solution was sonicated for 20 min to ensure complete dissolution of the DDEA and PLUM. The prepared solution was then analysed spectrophotometrically.

**3 Result and Discussion**

The developed method was validated in accordance with ICH Q2B guidelines.

**3.1 Validation of developed method**

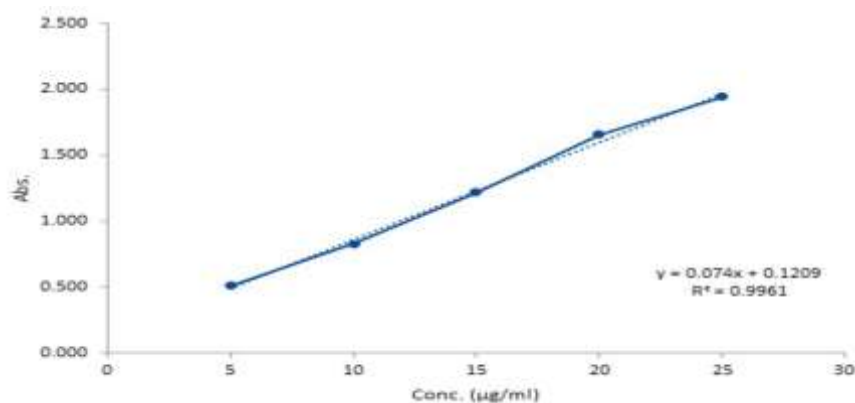
**3.1.1 Linearity**

DDEA and PLUM showed linear response in concentration range of 5-25 µg /ml and 5-40 µg /ml respectively, whereas linear regression equation for DDEA and PLUM was found to be  $y = 0.074x + 0.1209$  and  $y = 0.0212x + 0.0061$  with correlation regression of 0.9961 and 0.9987 respectively. Linearity table and calibration curve has been shown below in Table 1 and Fig.4.

**Table 1. Linearity of DDEA**

Concentration	Absorbance at 282 nm (n = 3)
5	0.5087
10	0.8280
15	1.2153
20	1.7553
25	1.9440

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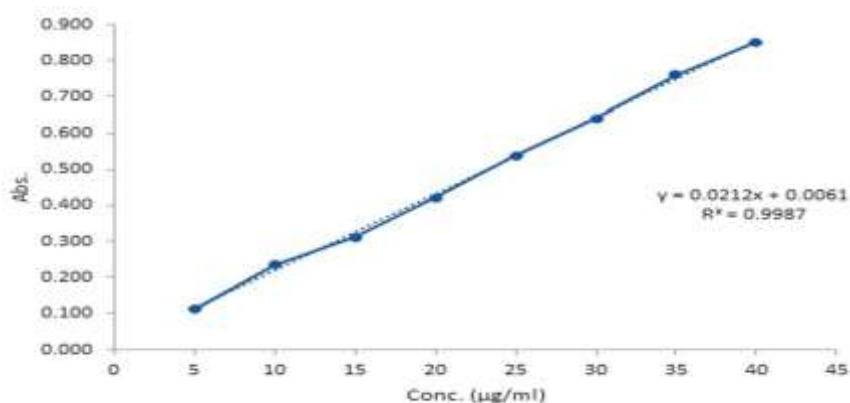
**Fig 4. Calibration curve of DDEA**

**Table 2. Linearity of PLUM**

Concentration	Absorption at 418 nm (n = 3)
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5	0.110
10	0.235
15	0.312
20	0.423
25	0.539
30	0.642
35	0.760
40	0.851



**Fig 5. Calibration curve of PLUM**

**3.1.2 Precision**

Repeatability studies, intraday precision as well as interday precision studies were carried out for DDEA and PLUM. The %RSD values were found to be less than 2%. Hence the method so developed was considered as precise (Table 3, 4, 5 and 6).

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**Table 3. Precision of DDEA (Intraday)**

Parameter	Absorption at 0 Hr	Absorption at 2 Hr	Absorption at 4 Hr	Absorption at 8 Hr
<b>DDEA Intraday (n = 6)</b>	1.3354	1.3215	1.3321	1.3215
	1.3242	1.3321	1.3451	1.3568
	1.3520	1.3345	1.3214	1.3211
	1.3440	1.3541	1.3542	1.3112
	1.3200	1.3420	1.3421	1.3211
	1.3452	1.3501	1.3554	1.3102



<b>Mean</b>	<b>1.3368</b>	<b>1.33905</b>	<b>1.341717</b>	<b>1.320983</b>
<b>SD</b>	<b>0.012621</b>	<b>0.012119</b>	<b>0.013114</b>	<b>0.011000</b>
<b>%RSD</b>	<b>0.944092</b>	<b>0.905023</b>	<b>0.977418</b>	<b>0.832705</b>

**Table 4. Precision of PLUM (Intraday)**

<b>Parameter</b>	<b>Absorption at 0 Hr</b>	<b>Absorption at 2 Hr</b>	<b>Absorption at 4 Hr</b>	<b>Absorption at 8 Hr</b>
	0.1847	0.1911	0.1889	0.1868
<b>PLUM</b>	0.1835	0.1875	0.1869	0.1861
<b>Intraday</b>	0.1876	0.1855	0.1852	0.1844
<b>(n = 6)</b>	0.1901	0.1868	0.1841	0.1834
	0.1865	0.1846	0.1849	0.1865
	0.1871	0.1888	0.1871	0.1878
<b>Mean</b>	<b>0.186583</b>	<b>0.187383</b>	<b>0.186183</b>	<b>0.185833</b>
<b>SD</b>	<b>0.002312</b>	<b>0.002344</b>	<b>0.001771</b>	<b>0.001631</b>
<b>%RSD</b>	<b>1.239162</b>	<b>1.250835</b>	<b>0.951399</b>	<b>0.877422</b>

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**Table 5. Precision of DDEA (Interday)**

<b>Parameter</b>	<b>Absorption in Day 1</b>	<b>Absorption in Day 2</b>	<b>Absorption in Day 3</b>
	1.3354	1.3201	1.3136
<b>DDEA</b>	1.3242	1.3101	1.2994
<b>Interday</b>	1.3520	1.3342	1.3216
<b>(n = 6)</b>	1.3440	1.3226	1.3147
	1.3200	1.3145	1.3201
	1.3452	1.3232	1.3105
<b>Mean</b>	<b>1.3368</b>	<b>1.320783</b>	<b>1.313317</b>
<b>SD</b>	<b>0.012621</b>	<b>0.008285</b>	<b>0.007978</b>
<b>%RSD</b>	<b>0.944092</b>	<b>0.627300</b>	<b>0.607456</b>

**Table 6. Precision of PLUM (Interday)**

<b>Parameter</b>	<b>Absorbance in Day 1</b>	<b>Absorption in Day 2</b>	<b>Absorption in Day 3</b>
	0.1847	0.1837	0.1866
	0.1835	0.1842	0.1827
<b>PLUM</b>	0.1876	0.1816	0.1854
<b>Interday</b>	0.1901	0.1839	0.1894
<b>(n = 6)</b>	0.1865	0.1857	0.1828
	0.1871	0.1828	0.1841
<b>Mean</b>	<b>0.186583</b>	<b>0.183650</b>	<b>0.184000</b>
<b>SD</b>	<b>0.002312</b>	<b>0.001378</b>	<b>0.001684</b>
<b>%RSD</b>	<b>1.239162</b>	<b>0.750363</b>	<b>0.915241</b>



### 3.1.3 Accuracy

For DDEA, the % mean recovery was found to be 99.75%, 98.98% and 98.54% , whereas for PLUM, 98.44 % , 98.00% and 99.98% at 80%, 100% and 120% levels respectively. The % RSD was found to be less than 2% which indicated the accuracy of method.

**Table 7. Recovery of DDEA**

Level	Set No.	% Recovery	%Mean Recovery	SD	%RSD
80%	1.	99.81	99.75	1.72	1.72
	2.	101.5			
	3.	98.06			
100%	1.	98.85	98.98	0.16	0.16
	2.	99.15			
	3.	98.9			
120%	1.	98.75	98.54	0.23	0.24
	2.	98.29			
	3.	98.62			

**Table 8. Recovery of PLUM**

Level	Set No.	% Recovery	%Mean Recovery	SD	%RSD
80%	1.	98.45	98.44	0.01	0.01
	2.	98.42			
	3.	98.44			
100%	1.	97.96	98.00	0.22	0.22
	2.	98.13			
	3.	98.03			
120%	1.	100.05	99.98	0.19	0.19

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### 3.1.4 LOD and LOQ

LOD and LOQ were measured using response and slope of regression. The LOD was found to be 0.80 µg/ml and 2.43 µg/ml and LOQ to be 1.24 µg/ml and 3.77 µg/ml for DDEA and PLUM respectively.

### 3.1.5 Robustness

Wavelength was altered and checked for the robustness of method. Results obtained showed %RSD less than 2% indicating robustness of developed method.

**Table 9. Robustness of DDEA**

Parameter	Absorbance at 281nm	Absorption at 282nm	Absorption at 283nm
	1.3254	1.3201	1.3236
<b>DDEA</b>	1.3142	1.3101	1.3194
<b>(n = 6)</b>	1.3239	1.3242	1.3216
	1.3240	1.3226	1.3147
	1.3201	1.3192	1.3201
	1.3250	1.3232	1.3105
<b>Mean</b>	<b>1.322</b>	<b>1.320</b>	<b>1.318</b>
<b>SD</b>	<b>0.004</b>	<b>0.005</b>	<b>0.005</b>
<b>%RSD</b>	<b>0.326</b>	<b>0.391</b>	<b>0.367</b>





**Table 10. Robustness of PLUM**

Parameter	Absorbance at 417nm	Absorption at 418nm	Absorption at 419nm
	0.1842	0.1837	0.1841
	0.1838	0.1842	0.1839
<b>PLUM</b>	0.1812	0.1816	0.181
<b>(n = 6)</b>	0.1841	0.1839	0.1832
	0.1865	0.1857	0.1862
	0.1822	0.1828	0.1826
<b>Mean</b>	<b>0.184</b>	<b>0.184</b>	<b>0.184</b>
<b>SD</b>	<b>0.002</b>	<b>0.001</b>	<b>0.002</b>
<b>%RSD</b>	<b>0.997</b>	<b>0.750</b>	<b>0.943</b>

### 3.2 Assay of gel formulation

The drug content of the formulation was found to be 99.84% and 99.33% for DDEA and PLUM respectively.

**Table 11. Drug content of formulation**

S.No.	% Drug content	
	DDEA	PLUM
1.	100.09	99.13
2.	99.77	99.53
3.	99.68	99.35
<b>Mean</b>	99.84	99.33
<b>S.D.</b>	0.215	0.200
<b>%RSD</b>	0.215	0.201

## 4 Conclusion

A simultaneous equation UV spectrophotometry method for the detection of DDEA and PLUM in combination dosage form was developed and validated in accordance with ICH guidelines. From the aforementioned techniques, a linearity of curve was found, indicating that the solution or dilution followed Beer's law between 5 and 25 µg/ml (DDEA and PLUM). The advantages of the proposed methodology for analytical purposes included quick results, cost effectiveness, simple preparation, good reproducibility, accuracy, and practicality.

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